



SEVA BHARATI MAHAVIDYALAYA

KAPGARI, JHARGRAM

DEPARTMENT OF CHEMISTRY

Program Outcomes, Program Specific Outcomes and Course Outcomes

Department of Chemistry

Programme Outcomes: B. Sc. Chemistry (General)

Department of Chemistry	After successful completion of three year degree program in Chemistry a student should be able to;
Programme Outcomes	<p>PO-1. Demonstrate, solve and an understanding of major concepts in all disciplines of chemistry.</p> <p>PO-2. Solve the problem and also think methodically, independently and draw a logical conclusion.</p> <p>PO-3. Employ critical thinking and the scientific knowledge to design, carry out, record and analyse the results of chemical reactions.</p> <p>PO-4. Create an awareness of the impact of chemistry on the environment, society, and development outside the scientific community.</p> <p>PO-5. Find out the green route for chemical reaction for sustainable development.</p> <p>PO-6. To inculcate the scientific temperament in the students and outside the scientific community.</p> <p>PO-7. Use modern techniques, various equipments and Chemical softwares.</p>

Programme Specific Outcomes	<p>PSO-1. Gain the knowledge of Chemistry through theory and practical experiments.</p> <p>PSO-2. To explain nomenclature, stereochemistry, structures, reactivity, and mechanism of the chemical reactions.</p> <p>PSO-3. Identify chemical formulae and solve numerical problems.</p> <p>PSO-4. To understand the basic principles of Organic, Inorganic, Physical and Analytical Chemistry and its applications through Various laboratory experiments.</p> <p>PSO-5. Use modern chemical tools, Models, Chem-draw, Charts and Equipments.</p> <p>PSO-6. Understand good laboratory practices and safety.</p> <p>PSO-7. Develop research oriented skills.</p> <p>PSO-8. Aware and handle the sophisticated instruments/equipments.</p>
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Course Outcomes B. Sc Chemistry (General)

Semester - I

Course	Outcomes
	After completion of these courses students should be able to:
DSC-1A	<p>CO-1. Understanding the Atomic Structure.</p> <p>CO-2. Understanding chemical bonding and molecular structure.</p> <p>CO-3. Understanding the fundamentals of Organic Chemistry.</p> <p>CO-4. Understanding the stereo chemistry of Organic Chemistry.</p> <p>CO-5. Understanding the preparations and reactions of Aliphatic Hydrocarbons, Alkanes, Alkenes and Alkynes.</p>

SEMESTER – II

DSC-1B	<p>CO-1. Understanding Chemical Energetic. Review of Thermodynamics.</p> <p>CO-2. Understanding Chemical Equilibrium.</p> <p>CO-3. Understanding Ionic Equilibrium.</p> <p>CO-4. Understanding the preparations and reactions of Aromatic Hydrocarbons, Alkyl and Aryl Halides, Alcohols, Phenols, Ethers and Carbonyl Compounds.</p>
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SEMESTER – III

DSC-1C	<p>CO-1. Understanding the Ideal Solutions, River Rule, Principle of Steam Distillation and Nerst Distribution Law.</p> <p>CO-2. Understanding Phase Rule, Clausius-Clapeyron Equation and Eutectic Temperature.</p> <p>CO-3. Understanding the conductance of strong and weak electrolytes, Kohlrausch Law.</p> <p>CO-4. Understanding Electro Chemistry.</p> <p>CO-5. Understanding the preparations and reactions of Carboxylic Acids and their derivatives, Amines, Diazonium Salts, Amino Acids and Carbohydrates.</p>
SEC-1	<p>CO-1. Understanding the analysis of Soil.</p> <p>CO-2. Understanding the analysis of Water.</p> <p>CO-3. Understanding the analysis of Food Products.</p> <p>CO-4. Understanding the idea of Chromatography.</p> <p>CO-5. Understanding the analysis of Cosmetics.</p>
SEMESTER - IV	
DSC-1D	<p>CO-1. Understanding the idea of Coordination Chemistry.</p> <p>CO-2. Understand the idea of Crystal Field Theory.</p> <p>CO-3. Understanding the Kinetic Theory of Gases.</p> <p>CO-4. Understanding the idea of Liquids such as Surface tension and Viscosity of a liquid.</p> <p>CO-5. Understanding the Solid States.</p> <p>CO-6. Understanding the idea and explanations of Chemical Kinetics.</p>
SEC-2	<p>CO-1. Understanding the idea of Carbohydrates.</p> <p>CO-2. Understanding the Proteins, Enzymes, Lipides and Hormones.</p> <p>CO-3. Understanding the composition and functions of Blood coagulation and Urine.</p>
SEMESTER - V	
DSE-1A	<p>CO-1. Understanding Qualitative and Quantitative aspects of analysis.</p> <p>CO-2. Understanding Optical methods of analysis.</p> <p>CO-3. Understanding Thermal methods of analysis.</p> <p>CO-4. Understanding Electro-analytical methods.</p> <p>CO-5. Understanding Separation Techniques such as Solvent Extraction and Chromatographic methods.</p>

SEC-3	CO-1. Understanding drugs discovery and development. CO-2. Understanding the preparations methods of Aspirin, Paracetamol, Chloramphenicol etc . CO-3. Understanding Fermentation methods of Ethyle Alcohol, Citric acid, Vitamin B ¹² and Vitamin C.
SEMESTER - VI	
DSE-1B	CO-1. Understanding the idea of Green chemistry. CO-2. Understanding Twelve Principles of Green chemistry. CO-3. Understanding Green Synthesis. CO-4. Understanding the future trends in Green Chemistry.
SEC-4	CO-1. Understanding the idea of Natural and Synthetic Pesticides. CO-2. Understanding Organochlorines Pesticides and Organophosphates Pesticides.

